

REMARKS

Claims 1-17 are pending in the application. Claim 17 has been subjected to a restriction requirement. Claims 11 and 12 have been cancelled by this amendment. Accordingly, claims 1-10 and 13-17 are at issue.

The courteous interview granted by Examiner Ngo to applicant's undersigned attorney on January 14, 1998 is hereby acknowledged with appreciation. During the interview, the Office Action and proposed claim amendments were discussed.

With respect to the Office Action, it was stated that compounds of formula (II) of claim 17 are related to the compounds and process of the claims of elected Group I, and that the compounds of formula (II) would be considered as a part of Group I. Accordingly, applicant has amended claim 17 by deleting the compounds of formulae (III), (V), (VI), (VII), (VIII), and (X). Claim 17, therefore, should be considered on the merits at this time. In addition, claim 16 has been amended to delete processes (B) and (C) from claim 16.

The Office Action indicates that applicant has not filed a certified copy of GB 9401090.7, which applicant relies upon for a claim of foreign priority. However, it is submitted that applicant is not required to submit a certified copy of the priority document in this case.

The present application is the U.S. national phase application of PCT/EP95/00183. Accordingly, it is applicant's understanding that under the procedures of the PCT, a copy of the priority document will have been supplied to the U.S. Patent Office, pursuant to Rule 17 of the PCT regulations. Accordingly, it is requested that the next communication concerning this application

contains an indication that the appropriate priority document is in the file of this application.

The Office Action also requests that a reference to an earlier-filed, and copending, application should be added to the present application. It is submitted that such a reference is not needed because there was no previously filed application in the U.S. A PCT application, designating the U.S., was filed on January 19, 1995. Filing of the PCT application in the U.S. was perfected by entering the U.S. national phase on July 17, 1996. Accordingly, no reference to a previously filed and copending application is necessary.

Claims 1-16(A) stand rejected under 35 U.S.C. §112, first and second paragraphs, as being nonenabling or indefinite. In addition, claims 11 and 12 stand rejected under 35 U.S.C. §101. In view of the amendments to the claims, and for the reasons set forth below, it is submitted that these rejections have been overcome and should be withdrawn.

In particular, it is contended that the phrase "R¹ and R³ together represent . . . alkyl or alkenyl chain" is not clear. As indicated in the Office Action, this phrase is intended to recite that R¹ and R³ are taken together to form a ring. Accordingly, claim 1 has been amended to recite that R¹ and R³ are taken together as a component of a 5- or 6-membered ring. R¹ and R³, together, contribute three or four members of the ring, and the remaining two members are the nitrogen atom bonded to R¹ and the carbon atom bonded to R³.

From a reading of all substituents R¹ and R³ recited in claim 1, the phrase in question can only be construed as taking R¹ and R³ together to form a ring. For example, R¹ can be C₁₋₆ alkyl, C₂₋₆ alkenyl, or C₂₋₆ alkynyl (and others), and R³ can be hydrogen or C₁₋₃ alkyl.

Based on those definitions of R¹ and R³, the only reasonable construction of taking R¹ and R³ together is to form a ring. Any other construction would be redundant in view of the recited definitions of R¹ and R³. It is submitted, therefore, that the amendment to claim 1 overcomes this rejection under 35 U.S.C. §112, second paragraph.

The phrase "and salts and solvates" also is considered indefinite. Accordingly, applicant has amended claims 1, 2, 8, 9, and 10 to recite "or salts or solvates." This amendment clarifies that there are not multiple forms of the compound of formula (I) in one compound. It again is submitted that these amendments to claims 1, 2, 8, 9 and 10 overcome this rejection under 35 U.S.C. §112, second paragraph.

Claims 11 and 12 stand rejected under 35 U.S.C. §112, second paragraph, and 35 U.S.C. §101. In view of this amendment, which cancels claims 11 and 12, it is submitted that these rejections are now moot.

Claims 1-16(A) stand rejected under 35 U.S.C. §112, first paragraph, as not being enabling for a compound of formula (I) when R¹ and R³ are taken together form a chain, because such a compound cannot be synthesized. It is submitted that this contention is incorrect, and, for the reasons set forth below, this rejection should be withdrawn.

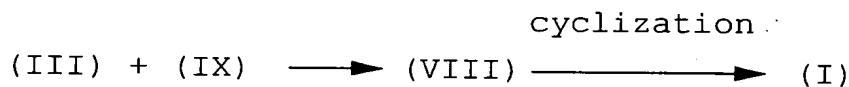
As stated above, when R¹ and R³ are taken together, these substituents form a ring. To illustrate how a compound of formula (I) can be made when R¹ and R³ are taken together to form a ring, the examiner's attention is directed to compound (IX) at page 13 of the specification. As stated in the specification,

"There is further provided by the present invention a process (B) for preparing a compound of formula (I), wherein R¹ and R³ together represent a 3- or 4-membered alkyl or alkenyl chain, which process (B) comprises cyclisation of a compound of formula (VIII)" (page 13, lines 3-6), and

"Conveniently a compound of formula (VIII) is prepared by reaction of a compound of formula (III) as hereinbefore described with a compound of formula (IX)" (page 13, lines 13-15).

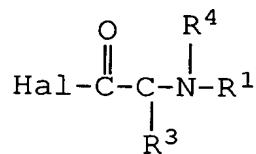
The compound of formula (III) is illustrated in the specification at the top of page 10.

Therefore, to prepare a compound of formula (I) wherein R¹ and R³ are taken together to form a ring, the following reaction sequence is used.

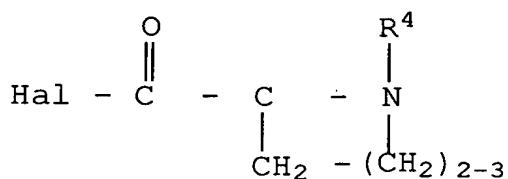


The compound of formula (III) is disclosed at page 10 of the specification. The cyclization step is disclosed at page 13, lines 8-12.

The synthesis of a compound of formula (I), wherein R¹ and R³ are taken together to form a ring, therefore, can be accomplished by providing a compound of formula (IX), wherein R¹ and R₃ are taken together to form a ring. In particular, a compound of formula (IX) has the following structure.

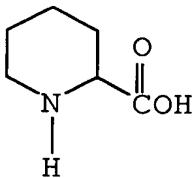


A compound of formula (IX) having R¹ and R³ taken together to form a 5- or 6-membered ring could have the structure

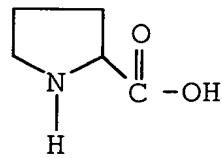


wherein Hal represent a halogen atom, like chlorine (see specification, page 9, lines 26-27), and R⁴ is a protecting group, like benzyloxycarbonyl (see specification, page 13, lines 17-18). Synthesis of the above compound would provide a compound of formula (IX), which in turn could be used to prepare a compound of formula (I) by the above reaction sequence.

The above compound can be prepared from compounds and synthetic steps well known to persons skilled in the art. For example, a suitable starting material would be picolinic acid or proline, having the following structures (a) and (b), respectively.



(a)

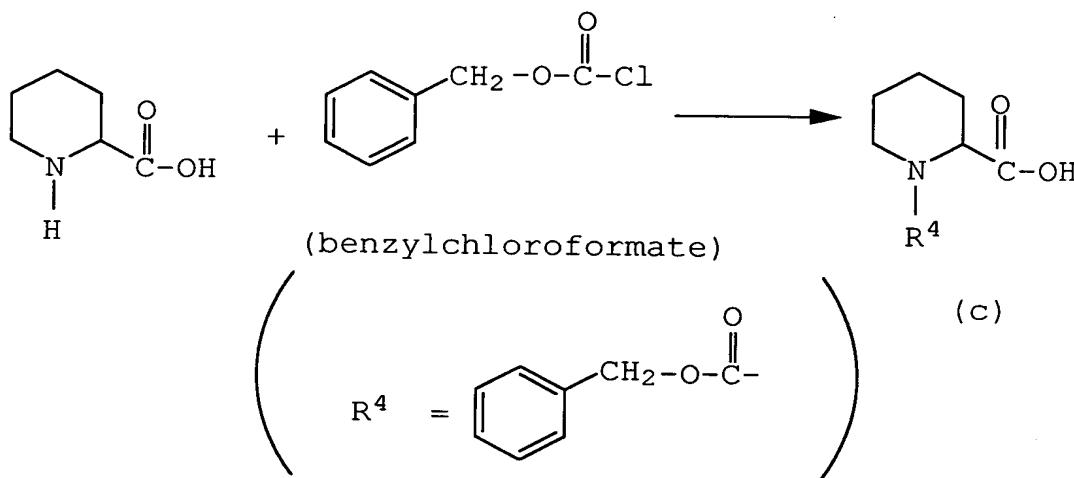


(b)

These compounds are available commercially, as illustrated in pages 1019 and 1057, of the 1992 catalog of

Aldrich Chemical Co., Milwaukee, WI, attached hereto as Exhibit A. In these starting materials, R¹ and R³ are taken together to form a 4-membered alkyl chain and a 3-membered alkyl chain, respectively, as a component of a 6-membered ring and a 5-membered ring, respectively.

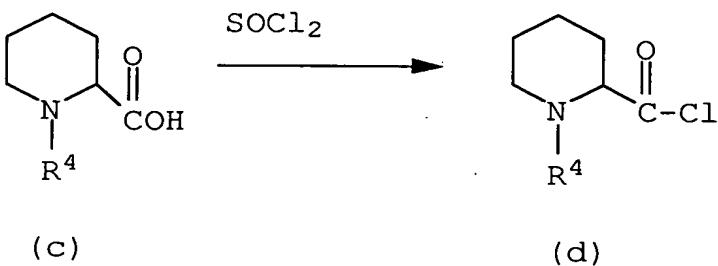
Compound (a) or (b) then can be reacted with a protecting compound, like benzylchloroformate, to position a protecting group, like benzyloxycarbonyl, on the nitrogen atom. This reaction is illustrated in C.D. Gutsche et al., "Fundamentals of Organic Chemistry," Prentice-Hall, Inc., Englewood Cliffs, NJ, page 1190, attached hereto as Exhibit B. This reaction is illustrated below for compound (a) and an identical reaction can be performed on compound (b).



Benzylchloroformate is available commercially, as illustrated at page 132 of Exhibit A.

Compound (c) then can be reacted with a common reagent for converting a carboxylic acid to an acid chloride, like thionyl chloride (SOCl₂), to provide com-

pound (d). This reaction is illustrated at page 39 of Exhibit B.



Compound (d) corresponds to the compound of formula (IX) at page 13 of the specification, wherein Hal is chlorine, R^4 is benzyloxycarbonyl, and R^1 and R^3 are taken together as a 4-membered alkyl chain component of a 6-membered ring. An identical reaction sequence starting with proline would yield an identical compound of formula (IX), except R^1 and R^3 are taken together as a 3-membered alkyl component of a 5-membered ring.

Compound (d), or a similar compound prepared from proline, then could be reacted with a compound of formula (III) to yield a compound of formula (VIII), which in turn is cyclized to form a compound of formula (I).

Therefore, a compound of formula (I) can be prepared when R^1 and R^3 are taken together as a 3- or 4-membered alkyl or alkenyl chain. The synthesis utilizes well-known starting materials, reagents, and reactions. Accordingly, it is submitted that the rejection of claims 1-16(A) under 35 U.S.C. §112, first paragraph, should be withdrawn.

During the interview, the examiner stated that the claims appeared excessive in scope because of the terms aryl and heteroaryl. In response, applicant has

amended claim 1 to more particularly claim the aryl and heteroaryl substituents. Support for this amendment can be found in the specification at page 2, lines 16-17.

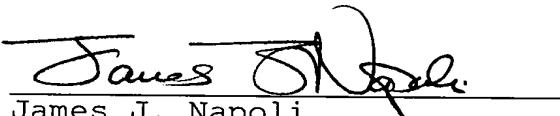
It is submitted that the claims are now in proper form and scope for allowance. Early and favorable action on the merits are respectfully requested.

Should the examiner wish to discuss the foregoing, or any matter of form in an effort to advance this application toward allowance, the examiner is urged to telephone the undersigned at the indicated number.

Respectfully submitted,

MARSHALL, O'TOOLE, GERSTEIN,
MURRAY & BORUN

By

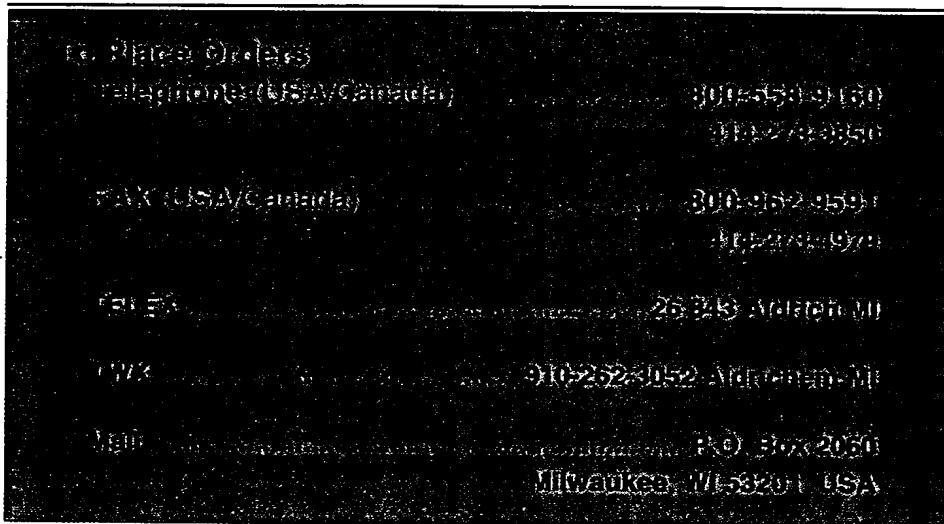

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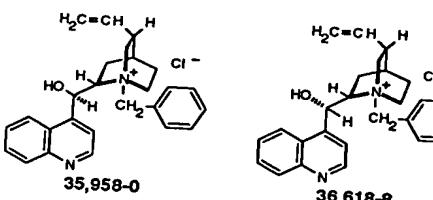
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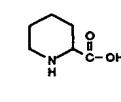
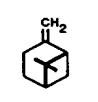
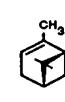
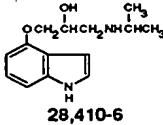
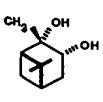
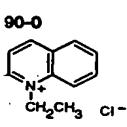
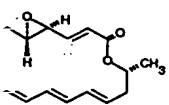
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Benzylbis(triphenylphosphine)palladium(II) chloride, see 27,766-5, <i>trans</i> -Benzyl-				\$
B1,790-5 Benzyl bromide , 98% [100-39-0] (α -bromotoluene) $C_6H_5CH_2Br$ FW 171.04.....	25g	9.40		
★ mp -3 to -1° bp 198-199° n_B 1.5750 d 1.438 Fp 188°F(86°C) <i>Bell.</i> 5,306 <i>Fleser</i> 5,25 <i>Merck Index</i> 11,1142 <i>NMR</i> 2(1),774D <i>FT-IR</i> 1(3),890B <i>Safety</i> 2,381B <i>RTECS# XS7965000 Disp. C CORROSIVE LACHRYMATOR</i>	100g	23.80		
24,563-1 Benzyl 2-bromoacetate , 96% [5437-45-6] $BrCH_2COCH_2C_6H_5$ FW 229.08.....	500g	77.95		
bp 166-170°/22mm n_B 1.5440 d 1.446 Fp >230°F(110°C) <i>Bell.</i> 6(1),220 <i>NMR</i> 2(2),269B <i>FT-IR</i> 1(3),1336D <i>Safety</i> 2,381C <i>RTECS# AF5957215 Disp. A IRRITANT</i>	50g	15.00		
38,204-3 Benzyl 3-bromopropyl ether [54314-84-0] $C_6H_5CH_2O(CH_2)_2Br$ FW 229.12.....	250g	48.60		
Disp. A IRRITANT	1g	8.00		
30,850-1 Benzyl-tert-butanol , see B1,800-6, α,α -Dimethylbenzenepropanol page 492	10g	62.00		
★ FW 312.37 n_B 1.5400 d 1.100 Fp >230°F(110°C) <i>Bell.</i> 9(2),594 <i>FT-IR</i> 1(3),1377A <i>Safety</i> 2,381D <i>RTECS# TH9990000 Disp. A IRRITANT</i>	5ml	10.00		
B1,820-0 Benzyl carbamate , 99% [621-84-1] $H_2NCOCH_2C_6H_5$ FW 151.17 mp 87-89°.....	250ml	14.60		
<i>Bell.</i> 6,437 <i>NMR</i> 2(2),361A <i>FT-IR</i> 1(2),389D <i>Disp. A IRRITANT</i>	1L	40.60		
12,101-0 S-Benzyl-N-carbobenzyloxy-L-cysteine , 98% [3257-18-9].....	25g	21.20		
$C_6H_5CH_2SCH_2CH(NHCO_2CH_2C_6H_5)COH$ FW 345.42 mp 94-96° $[\alpha]^{25}_D -44^\circ$ (c = 2, C_6H_5OH) <i>NMR</i> 2(2),264C <i>FT-IR</i> 1(2),265D <i>Disp. A IRRITANT</i>	100g	51.05		
1-Benzyl-3-carbomethoxy-4-piperidone hydrochloride, see 22,700-5, Methyl 1-benzyl-4-oxo-3-piperidinocarboxylate hydrochloride page 826	25g	13.05		
22,900-8 Benzylcetyltrimethylammonium chloride monohydrate	25g	27.95		
★ $C_6H_5CH_2N[(CH_3)_3CH_2]Cl \cdot H_2O$ FW 414.12 mp 62-64° <i>Bell.</i> 12(3),2212 <i>Merck Index</i> 11,2009 <i>NMR</i> 2(1),1122B <i>FT-IR</i> 1(1),1321C <i>Safety</i> 2,382A <i>RTECS# BO6822450 Disp. A CORROSIVE</i>	100g	8.80		
28,847-0 Benzyl-α-^{13}C chloride , 99 atom % ^{13}C [57742-41-3] (α -chlorotoluene- α - ^{13}C).....	250mg	13.20		
$C_6H_5^{\alpha-13}CH_2Cl$ FW 127.58 mp 43° bp 177-181° n_B 1.5380 d 1.100 Fp 165°F(73°C) <i>Safety</i> 2,382C <i>Disp. C HIGHLY TOXIC</i>	1g	184.10		
(Packaged in prescored ampules)	1g	455.80		
21,733-6 Benzyl-d₄ chloride , 99 + atom % D [59502-05-5] (α -chlorotoluene-d ₄).....	1g	82.60		
$C_6D_4CD_3Cl$ FW 133.64 bp 65°/10mm n_B 1.5374 d 1.200 Fp 165°F(73°C) <i>NMR</i> 2(1),6C <i>FT-IR</i> 1(3),1618C <i>Safety</i> 2,383A <i>Disp. C HIGHLY TOXIC</i> <i>CANCER SUSPECT AGENT</i>				
18,555-8 Benzyl chloride , 99% [100-44-7] (α -chlorotoluene) $C_6H_5CH_2Cl$ FW 126.59.....	50g	10.00		
★ mp -43° bp 177-181° n_B 1.5380 d 1.100 Fp 165°F(73°C) <i>Bell.</i> 5,292 <i>Merck Index</i> 11,1143 <i>NMR</i> 2(1),774C <i>FT-IR</i> 1(3),890A <i>Safety</i> 2,382B <i>RTECS# XS8925000 Disp. C HIGHLY TOXIC CANCER SUSPECT AGENT</i>	250g	12.90		
Mutagen, possible carcinogen. <i>Proc. Nat. Acad. Sci. U.S.A.</i> , 72, 979 (1975). Inhibited with 0.25% propylene oxide	1kg	17.90		
32,016-1 Benzyl chloride , 97% [100-44-7] (α -chlorotoluene) $C_6H_5CH_2Cl$	4kg†	48.20		
★ Inhibited with 0.25% propylene oxide (Packaged in poly-coated bottle)	1L	21.20		
13,359-0 Benzyl chloride [100-44-7] (α -chlorotoluene) $C_6H_5CH_2Cl$				
★ Inhibited with 0.25% propylene oxide	100g	11.90		
27,766-5 trans -Benzyl(chloro)bis(triphenylphosphine)palladium(II) [22784-59-4].....	1kg	17.40		
[benzylbis(triphenylphosphine)palladium(II) chloride] [(C_6H_5) ₂ Pd($CH_2C_6H_5$)Cl] 100mg FW 757.58 mp 166-170° <i>Fleser</i> 8,35 9,41 10,26 12,44 13,30 <i>Safety</i> 2,383C Catalyst for the coupling of alkyl-, vinyl- or alkynyltin groups with acyl chlorides. 1. <i>J. Am. Chem. Soc.</i> , 100, 3636 (1978). 2. <i>Tetrahedron Lett.</i> , 24, 2361 (1983). 3. <i>J. Org. Chem.</i> , 47, 2549 (1982). See also, <i>Aldrichimica Acta</i> , 17(3), 75 (1984).	4kg†	46.30		
11,993-8 Benzyl chloroformate , tech., 95% [501-53-1] (carbobenzoxy chloride).....	5g	11.90		
$CICO_2CH_2C_6H_5$ FW 170.60 n_B 1.5190 d 1.195 Fp 197°F(91°C) <i>Bell.</i> 8,437 <i>Fleser</i> 1,109 2,59 15,22 <i>Merck Index</i> 11,1807 <i>NMR</i> 2(2),331D <i>FT-IR</i> 1(2),353C <i>Safety</i> 2,383D <i>RTECS# LQ5860000 Disp. A HIGHLY TOXIC CANCER SUSPECT AGENT</i>	100g	28.40		
Protecting reagent in peptide synthesis. May contain <3% benzyl chloride	5x100g	86.10		

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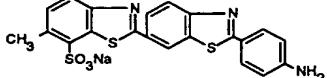


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enyl)....	1g 5g	0 20	8-7 (1S,2S,3R,5S)(+)-Pinanediol, 99% [18680-27-8] {[1S-(1 α ,2 α ,3 α ,5 α)]-2,6,6-trimethylbicyclo[3.1.1]heptane-2,3-diol} FW 170.25 mp 57-59° Fp >230°F(110°C) [α]D +8.5° (c = 6.5, C ₆ H ₅ CH ₃) Bell. 6(3),4145 FT-IR 1(3),250D Disp. A Chiral reagent in the synthesis of (2S,3S)- and (2R,3S)-3-phenyl-2-butanol. J. Am. Chem. Soc., 102, 7590 (1980).		1g 5g	13.70 45.60	
Merck... A	5g	45.5	3-Pinanol, see Isopinocampheol				
			10-6 Pinolol, 97% [13523-86-9] {1-(1H-indol-4-yloxy)-3-[(1-methylethyl)amino]-2-propanol} FW 248.33 mp 167-171° Merck Index 11,7412 Safety 2,2840C RTECS# UB6660000 Disp. A TOXIC IRRITANT		1g	11.80 26.90	
	5g 10g	49.4 61.0	107-0 (1R)(+)- α -Pinene, 99 + % [7785-70-8] FW 136.24 mp -62° bp 155-156° n _D 1.4660 d 0.857 Fp 90°F(32°C) [α]D ²⁵ +50.7° (neat) Bell. 5,146 Merck Index 11,7414 Safety 2,2841A Disp. D FLAMMABLE LIQUID IRRITANT Precursor to monoisopinocampheylborane (IPCBH ₃) used to prepare (+)-trans-2-methylcyclohexanol from 1-methylcyclohexene in 72.4% e.e. J. Am. Chem. Soc., 99, 5514 (1977). Chiral Intermediate. Aldrichimica Acta, 13(1), 13 (1980). Ibid., 20(1), 24 (1987). 98 + % e.e.		5g 25g	17.00 55.40	
	25g 100g	20.00 58.30	108-0 (1R)(+)- α -Pinene, 98% [7785-70-8] [α]D ²⁵ +47.1° (neat) 91 + % e.e.		100ml 500ml	22.40 77.00	
nm.....	5g 25g	21.80 70.30	1145-1 (1R)(+)- α -Pinene, tech., 85% [7785-70-8] [α]D ²⁵ +43° (neat) 91 + % e.e.		100ml 500ml	15.20 48.40	
			1752-4 (\pm)- α -Pinene, 98% [2437-95-8] FW 136.24 bp 155-156° n _D 1.4650 d 0.858 Fp 90°F(32°C) Bell. 5,144 Merck Index 11,7414 NMR 2(1),52D FT-IR 1(3),75B Safety 2,2840D Disp. D FLAMMABLE LIQUID IRRITANT		5ml 250ml 1L	11.90 22.80 62.20	
Merck... A	5g 100g 500g	16.80 49.75 167.50	10571-5 (1S)- α -Pinene, 99% [7785-26-4] FW 136.24 mp 155-156° n _D 1.4650 d 0.855 Fp 90°F(32°C) [α]D ²⁵ -50.7° (neat) Bell. 5,144 Merck Index 11,7414 Safety 2,2841B Disp. D FLAMMABLE LIQUID IRRITANT 98 + % e.e.		5g 25g	17.40 58.50	
	5ml 100ml 500ml	10.85 15.55 64.55	17439-9 (1S)(-)- α -Pinene, 99 + % [7785-26-4] [α]D ²⁵ -45° (neat) 87 + % e.e.		25g 100g 100ml 500ml	12.70 32.60 9.85 29.85	
			11208-9 (1S)(-)- α -Pinene, 99% [18172-67-3] FW 136.24 mp -61° bp 165-167° n _D 1.4780. d 0.859 Fp 91°F(32°C) [α]D ²⁵ -21° (neat) Bell. 5,154 Merck Index 11,7415 NMR 2(1),54A FT-IR 1(3),75D Safety 2,2841C RTECS# DT5077000 Disp. D FLAMMABLE LIQUID IRRITANT Chiral Intermediate. Aldrichimica Acta, 13(1), 13 (1980).		5ml 250ml 1L 4L	11.90 18.30 56.40 97.85	
	1g 5g	17.90 65.40	21830-8 α -Pinene oxide, 98% [1686-14-2] FW 152.24 bp 102-103°/50mm n _D 1.4690. d 0.964 Fp 150°F(65°C) [α]D ²⁵ -81° (neat) Bell. 5,152 NMR 2(1),198D FT-IR 1(3),312D Safety 2,2841D RTECS# RP5600000 Disp. C		50g 250g	18.50 60.00	
	1g	27.75	21831-6 β -Pinene oxide, 90% [6931-54-0] FW 152.24 bp 98-100°/27mm n _D 1.4770. d 0.976 Fp 151°F(66°C) [α]D ²⁵ +7° (neat) Bell. 17(2),44 NMR 2(1),199A FT-IR 1(1),237B Safety 2,2842A RTECS# TK4570000 Disp. C		50g	26.00	
cc.....	1g 5g	21.80 61.00	11,010-8 cis- α -Pinonic acid, 98% [473-72-3] (cis-3-acetyl-2,2-dimethyl-cyclobutaneacetic acid) CH ₃ COCH ₂ (CH ₃) ₂ CH ₂ CO ₂ H FW 184.24 mp 104-107° Bell. 10,622 NMR 2(1),469D FT-IR 1(1),533A Safety 2,2842B Disp. C IRRITANT		5g 25g	22.45 73.10	
	1g 5g	13.70 45.60	Pipecolic acid, see P4585-0, Pipecolinic acid page 1019				
			Pipecoline, see Methylpiperidine				
			20,806-2 D-Pipecolinic acid, 99% [1723-00-8] [(R)(+)-2-piperidinocarboxylic acid] FW 129.16 mp 27°(dec.) [α]D ²⁵ +27° (c = 1, H ₂ O) Bell. 22,8 Merck Index 11,7425 Safety 2,2842D Disp. A IRRITANT		25mg 100mg	22.10 59.85	
			P4,585-0 DL-Pipecolinic acid, 98% [4043-87-2] (2-piperidinocarboxylic acid) FW 129.16 mp 28°(dec.) Bell. 22,7 Merck Index 11,7425 NMR 2(1),505C FT-IR 1(1),585B Safety 2,2843A RTECS# TK6021000 Disp. A IRRITANT		25g 100g	41.30 112.00	
			24,852-5 DL-Pipecolinic acid hydrochloride, 99% [5107-10-8] (2-piperidine-carboxylic acid) FW 165.62 mp 263-266° Bell. 22,7 Merck Index 11,7425 FT-IR 1(1),585C Safety 2,2843B Disp. A IRRITANT HYGROSCOPIC		5g	28.95	
			23,775-2 L-Pipecolinic acid, 99% [3105-95-1] [(S)(-)-2-piperidinocarboxylic acid] FW 129.16 mp 27° [α]D ²⁵ -26.4° (c = 1, H ₂ O) Bell. 22,8 Merck Index 11,7425 NMR 2(1),505D FT-IR 1(1),585D Safety 2,2842C Disp. A IRRITANT Proline homolog. Occurs in seeds, malt, edible mushrooms, fruits, etc.		100mg	23.60	

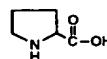
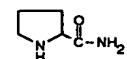


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FW 140.81 Merck... ID	10g 5g 25g	27.20 97.60 24.80 84.90	1702-4 4,768-4 16,990-2	5β-Pregnane-3α,17α,20α-triol, 98% [1098-45-9] FW 338.52 mp 250-252°. [α] _D ²⁵ -3.5° (c = 0.5, C ₂ H ₅ OH) Bell. 8(3),6405 Safety 2,2926D Disp. A Pregnenolone, 98% [145-13-1] FW 316.49 mp 190-192°. [α] _D ²⁵ + 27° (c = 1, C ₂ H ₅ OH) Merck Index 11,7739 NMR 2(2),923A FT-IR 1(2),1051D Safety 2,2940C RTECS# TU5560700 Disp. A	25mg 100mg 5g 25g	12.40 31.90 11.20 35.60	
1.010 Fp none.....	100ml 500ml	14.00 36.00	16,990-2	Pregnenolone acetate, 99% [1778-02-5] FW 358.52 mp 149-152°. [α] _D ²⁵ + 19° (c = 1, C ₂ H ₅ OH) Merck Index 11,7739 NMR 2(2),936D FT-IR 1(2),1061C Safety 2,2940D Disp. A	5g 25g	9.45 33.30	
none Disp. H.....	100ml 250ml	78.10 125.40		Prehnitene, see 15,360-5, 1,2,3,4-Tetramethylbenzene page 1171			
concentration				Prenyl bromide, see 24,990-4 4-Bromo-2-methyl-2-butene page 198			
I ₂ O FW 318.04.....	25g 100g	18.10 51.60		Pr(fod) _n , see 16,135-7, Resolve-AI PrFOD® page 1088			
8] PrCl ₃ ·7H ₂ O.....	5g	46.20		Pr(hfc) _n , see Tris[3-(heptafluoropropylhydroxymethylene)camphorato], praseodymium(III) derivative			
Q IRRITANT							
PrCl ₃ ·6H ₂ O.....	50g 5000 Disp. Q	32.40 122.50	16,039-3	Primaquine diphosphate, 99% [63-45-6] [8-(4-amino-1-methylbutylamino)-6-methoxyquinoline] FW 455.35 mp 205-206° (dec.) Merck Index 11,7751 NMR 2(2),741A FT-IR 1(2),864B Safety 2,2942C RTECS# VA9660000 Disp. A TOXIC	1g 10g	7.10 35.55	
77-0] Pr(NO ₃) ₃ ·6H ₂ O.....	5g 25g	42.65 168.30	16,686-5	Primuline [8064-60-6] (C.I. 49000, Direct Yellow 59) FW 475.55 λ _{max} 356nm FT-IR 1(2),1039B UV-VIS 588 RTECS# TV1050000 Disp. A Useful in a simple retrograde double-labeling procedure for studying axonal branching, in combination with 4',6-diamidino-2-phenylindole dihydrochloride (DAPI, 21,708-5) and Evans Blue (20,633-4). Science, 204, 873 (1979). Dye content ~75%	25g 100g	13.60 42.10	
2] Pr(NO ₃) ₃ ·6H ₂ O.....	50g 250g	26.80 97.25	22,296-8	Pristane, see T2280-2, 2,6,10,14-Tetramethylpentadecane page 1175	25g	8.80	
7(C ₂ O ₄) ₂ ·xH ₂ O.....	25g 100g	17.10 48.40	24,303-5	Procainamide hydrochloride, 99% [614-39-1] [4-amino-N-(2-diethylaminoethyl)benzamide] H ₃ NC ₆ H ₄ CONHCH ₂ CH ₂ N(C ₂ H ₅) ₂ ·HCl FW 271.79 mp 167-169° Bell. 14(3),1077 Merck Index 11,7762 NMR 2(2),350C FT-IR 1(2),373B Safety 2,2942D RTECS# CV2295000 Disp. A IRRITANT	25g 100g	11.90 31.60	
FW 1021.44 Disp. O.	2g 10g	19.50 74.70	22,297-6	Procaine, 99 + % [59-46-1] [2-(diethylamino)ethyl 4-aminobenzoate] H ₃ NC ₆ H ₄ CO ₂ CH ₂ CH ₂ N(C ₂ H ₅) ₂ FW 236.32 mp 61-62° Bell. 14,424 Merck Index 11,7763 NMR 2(2),288B FT-IR 1(2),303C Safety 2,2943A RTECS# DG2100000 Disp. A TOXIC IRRITANT	25g 100g	11.90 31.60	
.....	50g 250g	39.40 118.00	22,297-6	Procaine hydrochloride, 99% [51-05-8] [2-(diethylamino)ethyl 4-aminobenzoate] H ₃ NC ₆ H ₄ CO ₂ CH ₂ CH ₂ N(C ₂ H ₅) ₂ ·HCl FW 272.78 mp 155-156° Bell. 14,424 NMR 2(2),349D FT-IR 1(2),303D Safety 2,2943B RTECS# DG2275000 Disp. A TOXIC IRRITANT	5g 100g	8.10 14.85	
FW 570.00 Merck.....	50g	27.20	27,255-2	Procion Blue HB, see 24,222-5, Reactive Blue 2 page 1085			
00 d 0.868.....	782g 6x782g 3.1kg 4x3.1kg 15.6kg	13.00 73.50 36.85 139.00 173.20	27,255-2	Procion Yellow H-E3G [59112-78-6] (Reactive Yellow 81) Disp. A	10g 50g	10.00 33.50	
chromene).....	1g 5g	29.80 106.60		Proflavine hemisulfate, see 19,822-6, 3,6-Diaminoacridine hemisulfate page 381			
F(110°C) Merck A				Proflavine hydrochloride, see 13,110-5, 3,6-Diaminoacridine hydrochloride page 381			
which induce trial insecticides.							
s, 54(16), 19 (1976).							
-3-chromene).....	250mg 1,7716	21.60 53.30	85,045-4	Progesterone, 98% [57-83-0] FW 314.47 mp 129-130° [α] _D ²⁵ + 182° (c = 2, dioxane) Merck Index 11,7783 FT-IR 1(2),1052B Safety 2,2944C RTECS# TW0175000 Disp. A CANCER SUSPECT AGENT MUTAGEN	5g 25g	9.30 36.20	
hodoros parkeri.							
a-1,4-diene-3,20-.....	1g 5g	9.10 37.40	28,705-9	L-Proline, 98% [7531-52-4] FW 114.15 mp 95-97° [α] _D ²⁵ -100° (c = 2, C ₂ H ₅ OH) Bell. 22(3),15 Disp. A	250mg 1g	11.80 31.80	
ne) Bell. 8(4),3467 Disp. A				28,891-9	100mg 500mg	6.30 17.90	
iene-3,11,20-trione)....	1g 5g	9.85 37.60		D-Proline, 99 + % [344-25-2] [(R)-(+)proline] FW 115.13 mp 223° (dec.) [α] _D ²⁵ + 85.0° (c = 4, H ₂ O) Bell. 22,2 Fieser 7,307, 9,393 FT-IR 1(1),583B Safety 2,2946D Disp. A	5g	77.10	
8(4),3531 Merck A				17,182-4	1g 5g	6.40 19.85	
esterone page 375				L-Proline, 99% [609-36-9] FW 115.13 mp 208° (dec.) Bell. 22,4 Fieser 9,393 Merck Index 11,7780 NMR 2(1),504A FT-IR 1(1),583A Safety 2,2946C Disp. A	2.5g 100g	5.90 33.50	
α,20α-diol).....	1mg 5mg	15.80 50.45	13,154-7	L-Proline, 99 + % [147-85-3] [(S)-(+)proline] FW 115.13 mp 228° (dec.) [α] _D ²⁵ -84° (c = 4, H ₂ O) Bell. 22,2 Fieser 8,492, 8,421, 9,393, 10,331, 12,414 Merck Index 11,7790 FT-IR 1(1),583C Safety 2,2946A RTECS# TW3584000 Disp. A Optically active intermediate for organic synthesis. Aldrichimica Acta, 13(1), 13 (1980).	2.5g 25g 100g	5.90 10.55 33.50	
γ-β-pregnane).....	100mg 500mg	17.80 61.55					
ck Index 11,7732 TU4157113 Disp. A							
19,491-3							



20,686-5 28,705-9



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**Fundamentals
of
Organic Chemistry**

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Acid halides are distinguished from the corresponding acids by a carbonyl stretching band at a significantly higher frequency in the infrared. Whereas type AC carboxylic acids absorb at $1725\text{--}1700\text{ cm}^{-1}$, the corresponding acid chlorides absorb at $1815\text{--}1770\text{ cm}^{-1}$ (see Table 15.2 on p. 394).

15.1c. SYNTHESIS OF ACID HALIDES. Acid halides are almost invariably synthesized from the corresponding carboxylic acids by the action of any one of several inorganic reagents, including phosphorus trichloride (PCl_3), phosphorus pentachloride (PCl_5), and thionyl chloride (SOCl_2) (Fig. 15.1).

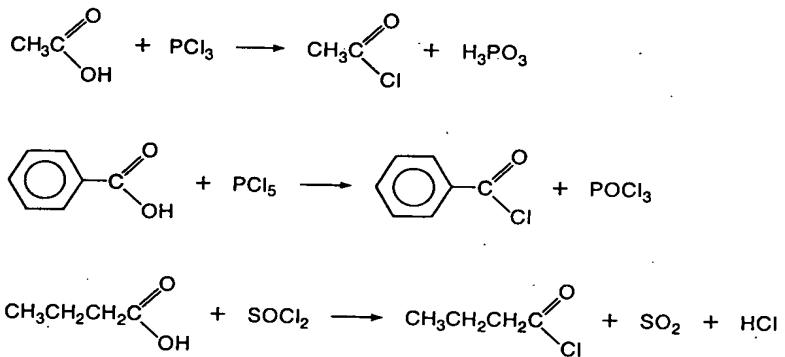


Fig. 15.1. Synthesis of acid chlorides from carboxylic acids.

15.1d. REACTIONS OF ACID HALIDES. We might predict that the electrophilic character of the carbonyl carbon in acid halides should be greater than in aldehydes and ketones because of the electron-withdrawing effect (*i.e.*, inductive effect) of the halogen atom. This does, in fact, prove to be the case, although to some extent this effect is counterbalanced by the electron-releasing effect of the nonbonded electrons of the halogen [*e.g.*, resonance structure Fig. 15.2(c)];

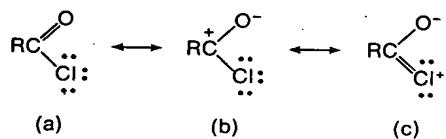
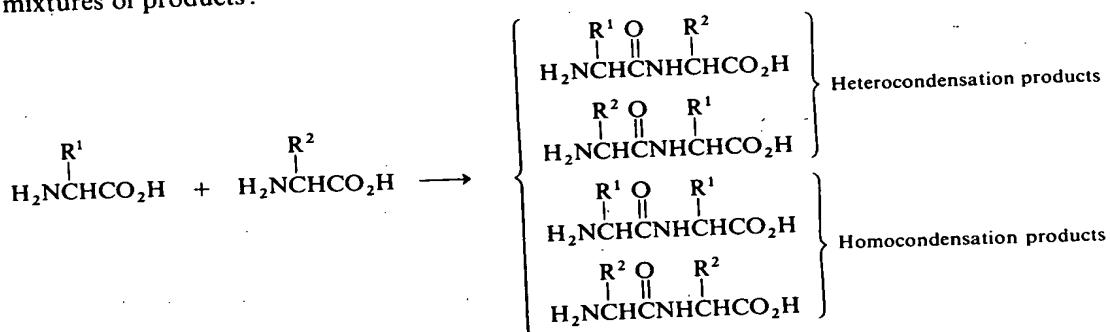


Fig. 15.2. Resonance structures of acid chlorides.

acid halides are exceedingly reactive toward nucleophilic reagents. The products of reaction, however, are different from those from aldehydes and ketones (see Section 13.5), for the initial step, producing an addition product, is succeeded by a second step, in which the halogen is eliminated and the carbonyl group is regenerated. The overall reaction is a *substitution* process, which proceeds via nucleophilic addition followed by elimination. The hydrolysis of acetyl chloride, for instance, can be depicted in this fashion (Fig. 15.3).

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First, amino acids are bifunctional molecules capable of undergoing facile reaction at the carboxyl function as well as the amino function. Thus, if we wish to form a peptide link between two *different* amino acids, we immediately face the problem associated with any mixed condensation—*viz.*, the formation of mixtures of products:



To circumvent this problem, we must "protect" or "block" the amino group of one of the participants and the carboxyl group of the other. Amino groups are frequently protected by means of the *t*-butoxycarbonyl (*t*-BOC) function, prepared by the action of *t*-butyl azidoformate [$(\text{CH}_3)_3\text{COCON}_3$] or *t*-butyl chloroformate [$(\text{CH}_3)_3\text{COOCOCl}$] on the amino acid. Carboxyl groups can be protected by conversion to the benzyl or *t*-butyl ester. All of these are good protecting groups because they are easily removed under mild acid-catalyzed hydrolysis (Fig. 39.9).

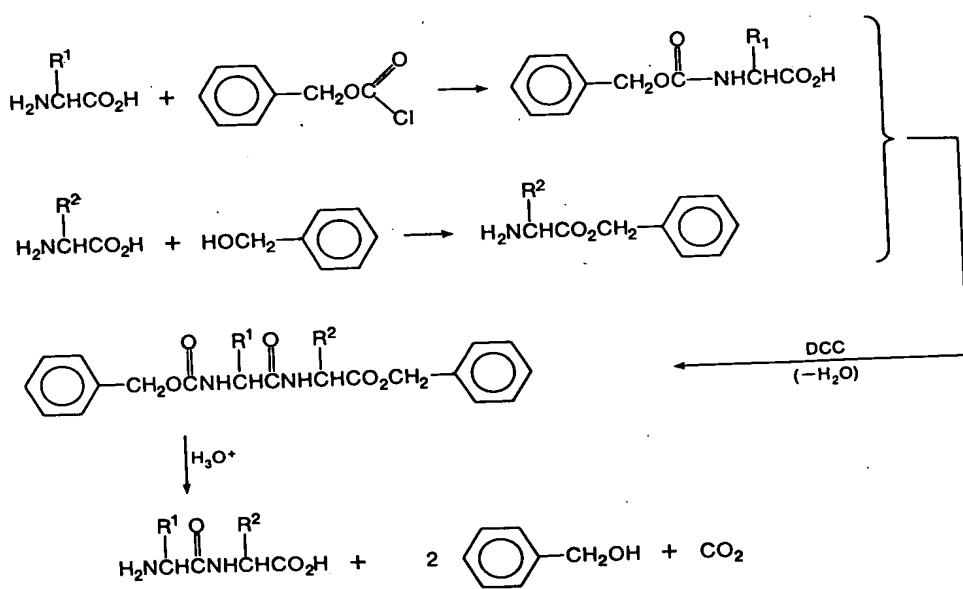


Fig. 39.9. Use of "protecting" groups in the synthesis of a dipeptide.

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